

Propolis from the Mediterranean Region: Chemical Composition and Antimicrobial Activity

Milena Velikova^a, Vassya Bankova^{*,a}, Kadriye Sorkun^b, Saadi Houcine^c, Iva Tsvetkova^d, A. Kujumgiev^d

- ^a Institute of Organic Chemistry with Centre of Phytochemistry, Bulgarian Academy of Sciences, 1113 Sofia, Bulgaria
- ^b Hacettepe University, Science Faculty, Dept. of Biology, Ankara, Turkey
- ^c Centre Universitaire Med Boudiat, M'Sila, Algeria
- ^d Institute of Microbiology, Bulgarian Academy of Sciences, 1113 Sofia, Bulgaria

* Author for correspondence and reprint requests

Z. Naturforsch. 55c, 790–793 (2000); received May 10/June 19, 2000

Propolis, *Populus*, Antimicrobial Activity

The chemical composition of propolis from Bulgaria, Turkey, Greece and Algeria was investigated by GC-MS. All of them contained mainly flavonoids and esters of caffeic and ferulic acids, which indicated that their main source are buds of poplars of the taxonomic section *Aegieros*. Some Turkish samples contained a low percent of diterpenic acids, while in Algerian samples significant amounts of a hydroxyditerpenic acid (M=322, its structure not determined by its MS) were found. All samples showed significant antibacterial and weak to moderate antifungal activity.

Introduction

Propolis (bee glue) is well known for its valuable biological activities (antibacterial, antifungal, immunostimulating, antitumor, antiinflammatory, etc.) (Marcucci, 1995; Burdock, 1998). Its chemical composition is variable, depending on the site of collection, because in different ecosystems different plant exudates and secretions could serve as a source of propolis. Numerous investigations have established that in Central Europe, as well as in North America and the non-tropical regions of Asia, the main propolis source is the exudate of poplar buds of *Populus* spp. (Greenaway *et al.*, 1990; Bankova *et al.*, 2000). These exudates contain typical phenolic compounds: flavanones and esters of caffeic and ferulic acids. In “border regions”, however, (Sonoran Desert, Tunisia, Egypt) some other plants interfere and new components appear in propolis (Wollenweber and Buchmann, 1997; Martos *et al.*, 1997; Christov *et al.*, 1998). The objective of this work is to study the chemical composition of bee glue from the Mediterranean region in order to find out the importance of poplar trees and other plants as its source. This information would be of importance for a future elaboration of propolis standards, for this reason antibacterial and antifungal tests were also carried out.

Experimental

Propolis samples

The geographic origin of the samples and time of collection are listed in Table I.

Extraction and sample preparation

Propolis, grated after cooling, was extracted for 24h with 70% EtOH at room temperature (1:10, w/v). The extract was evaporated to dryness. About 5 mg of the residue were mixed with 75 µl of dry pyridine and 50 µl bis(trimethylsilyl)trifluoracetamide (BSTFA), heated at 80 °C for 20 min and analysed by GC-MS.

Table I. Geographical origin and time of collection of propolis samples.

Sample	Location	Year
LOV	Lovetch, Bulgaria	1999
BUR	Aitos, Bulgaria	1999
GR	Nigrita, Greece	1997
BU-7	Bursa, Turkey	1997
BU-8	Bursa, Turkey	1998
MU	Mugla, Turkey	1998
IZ	Izmir, Turkey	1998
AN	Beytepe, Turkey	1998
AL-1	M'Sila, Algeria	1996
AL-2	M'Sila, Algeria	1998



GC-MS analysis

The GC-MS analysis was performed with a Hewlett Packard Gas Chromatograph 5890 Series II Plus linked to Hewlett Packard 5972 mass spectrometer system equipped with a 30 m × 0.25 mm ID SPB-1 column, film thickness 25 µm. A fused silica capillary column was used, mass selective detector, with He as carrier gas, linear velocity 35 cm/min, split ratio 1:50, temperature program 100–300 °C at 5 °C/min, injector temperature 300 °C.

Identification of compounds

The identification was accomplished using computer searches on commercial libraries. In some cases, when identical spectra have not been found, only the structural type of the corresponding component was proposed on the basis of its mass-spectral fragmentation. Reference compounds were co-chromatographed when possible to confirm GC retention times.

Antibacterial activity

For the investigation of the antibacterial activity the agar cup method was used with *Staphylococcus aureus* 209 and *Escherichia coli* WF+ as test strains. An inhibitory zone with a diameter less than 10 mm corresponds to lack of activity (10 mm is the diameter of the cup), 0.1 ml of test solution containing 0.4 mg propolis extract in ethanol was applied to every cup.

Antifungal activity

For the investigation of the antifungal activity the agar cup method was used (Spooner and Sykes, 1972). As test microorganism, *Candida albicans* 562 was used. The antifungal activity was measured as diameter of the inhibitory zones. An inhibitory zone with a diameter less than 10 mm corresponds to lack of activity (10 mm is the diameter of the agar cup), 0.1 ml of test solution containing 0.5 mg propolis extract in ethanol was applied to every cup. Control experiments with ethanol showed that solvent does not have any activity.

Results and Discussion

The chemical composition of two Bulgarian, one Greek, five Turkish and two Algerian samples

were investigated by GC-MS after silylation. The results obtained are presented in Table II.

All samples display the typical pattern of “poplar type” propolis. They contain the combination of secondary metabolites characteristic for the buds of *Populus* spp. of the section *Aigeiros*: pinocembrin, pinobanksin and its acetate, prenyl esters of caffeic and ferulic acids (Greenaway *et al.*, 1990; Wollenweber and Buchmann, 1997). The poplar origin of Bulgarian bee glue is well documented (Bankova *et al.*, 1992). Obviously the main source of the other samples investigated were also poplar buds. This is an expected result as far as the locations where propolis was collected are within the area of poplar species. On the other hand, several differences between the samples are evident. Some of these differences are quantitative and this could be explained by the well known fact that each species and even each clone of poplar has its own characteristic mixture of compounds in its bud exudate (Greenaway *et al.*, 1990).

More important are qualitative differences: samples from Algeria and Turkey contained diterpenic acids. Till now, diterpenic acids have only been found in Brazilian propolis (Bankova *et al.*, 1996; Banskota *et al.*, 1998). In two Turkish samples small amounts of pimaric and isopimaric (sample from Mugla); abietic and dihydroabietic acid (sample from Beytepe) were detected. In Algerian samples however, diterpenic acids appear to be among the main components, and especially a compound with M=322 (hydroxyditerpenic acid). Its structure could not be determined by its MS only. The plant source of these compounds remains unknown and is a question of future investigations.

These chemical differences, however, did not result in a significant change of the biological activity of the samples investigated. All samples showed good effect against *S. aureus*. The activity against *E. coli* was weak or lacking, which corresponds to the literature data (Marcucci, 1995). This is a confirmation that bees are able to find plant sources that provide a good defense against infections in any ecosystem they inhabit, as it was shown for Brazilian propolis (Kujumgiev *et al.*, 1999).

These results confirm that European propolis is definitely of poplar origin and this could be a key for its standardization based on quantification of

Table II. Chemical composition (%)^a of ethanol extracts of propolis.

Substance	LOV	BUR	GR	BU-7	BU-8	MU	IZ	AN	AL-1	AL-2
Aromatic acids										
Benzoic acid	0.3	0.4	0.3	3.9	1.4	0.2	0.1	0.8	0.4	0.4
<i>p</i> -Hydroxybenzoic acid	–	–	–	0.1	–	–	–	–	–	–
Vanillic acid	–	–	–	0.2	–	–	–	–	–	–
Cinnamic acid	–	–	–	0.1	–	1.1	0.1	0.2	–	–
<i>p</i> -Coumaric acid	0.6	0.4	0.5	3.9	–	–	1.3	0.4	0.1	0.4
Ferulic acid	0.3	0.3	0.4	14.4	–	1.5	0.7	0.8	0.1	0.8
Isoferulic acid	0.4	0.4	0.4	1.4	–	–	0.2	–	–	–
Caffeic acid	2.9	1.9	1.7	1.2	1.2	0.3	1.7	1.4	0.3	0.7
3,4-dimethoxycinnamic acid	0.5	0.4	0.9	–	–	–	5.8	0.5	–	–
Fatty acids										
Hexadecanoic acid	0.4	1.4	0.5	1.0	1.0	–	2.7	0.4	1.0	0.8
Oleic acid	0.4	1.2	0.8	0.7	2.5	0.7	2.9	2.3	0.5	–
Octadecanoic acid	–	–	–	0.2	–	–	–	0.1	–	–
Esters										
Benzyl benzoate	–	–	–	0.5	–	–	–	–	–	–
Benzyl cinnamate	–	–	–	4.1	0.4	–	–	–	–	–
Isopent-3-enyl ferulate	1.5	0.9	1.6	–	0.8	–	–	0.7	0.4	–
3,3-dimethylallyl ferulate	0.8	1.8	1.8	–	3.4	–	–	1.0	0.1	–
Isopent-3-enyl caffeate	0.7	0.8	1.5	0.5	1.0	5.2	2.6	4.9	1.7	0.6
3,3-dimethylallyl caffeate	2.1	4.3	2.5	3.0	4.0	3.2	3.6	5.6	3.0	1.4
Benzyl <i>p</i> -coumarate	–	–	–	0.8	0.3	–	1	0.1	–	–
Benzyl ferulate	–	–	–	7.8	3.0	–	–	–	0.4	–
Benzyl caffeate	8.3	3.0	4.3	1.2	3.5	–	1.3	1.2	–	4.5
Phenethyl caffeate	2.6	2.8	4.8	0.3	4.0	0.8	1.1	1.9	1.7	4.1
Cinnamyl caffeate	–	–	–	–	–	–	–	1.0	–	–
Flavonoids										
Chrysin	7.3	2.8	4.9	0.7	5.7	–	1.3	7.0	0.4	5.6
Galangin	6.8	3.5	5.8	1.3	–	2.2	2.9	8.9	5.9	5.3
Sakuranetin	–	–	–	–	–	–	–	2.5	0.6	0.5
3-Methylgalangin	–	–	–	0.7	–	–	–	–	–	–
Pinostrobin	1.4	0.9	3.2	0.4	0.9	0.4	0.1	1.5	0.4	–
Pinocembrin	14.4	10.9	10.3	8.9	0.3	1.3	7.8	4.9	5.0	–
Pinobanksin	5.5	1.7	1.5	1.0	3.5	1.4	0.2	2.5	7.4	9.3
Pinobanksin 3-O-acetate	15.6	9.2	6.4	3.0	6.6	1.2	1.0	8.0	8.9	5.3
Pinobanksin 3-O-Butenoate	2.9	1.5	2.5	–	4.7	–	0.1	1.5	2.5	1.6
Pinobanksin 3-O-Hexanoate	2.0	1.5	1.4	–	0.7	–	0.9	2.3	0.7	1.2
Diterpenic acids										
Pimaric acid	–	–	–	0.2	–	–	–	0.2	1.2	0.8
Isopimaric acid	–	–	–	–	–	0.8	–	–	0.5	–
Abietic acid	–	–	–	–	–	1.0	–	–	–	–
Dehydroabietic	–	–	–	–	–	–	–	0.9	–	–
Diterpenic hydroxyacid M=322	–	–	–	–	–	–	–	–	12.4	21.1
Sugars										
Fructose	1.2	1.4	–	1.0	–	8.6	9.5	0.2	–	–
Glucose	0.4	0.6	0.5	1.5	–	5.7	17.1	0.2	–	–
Sucrose	–	–	–	–	–	3.9	–	–	–	–
Maltose	–	–	–	–	–	0.2	–	–	–	–
Others										
Glycerol	2.4	0.6	0.6	2.8	0.5	5.6	7.2	0.6	0.1	–
Vanillin	–	–	–	1.2	0.4	–	–	–	–	–
Mallic acid	–	–	–	–	–	1.2	–	0.2	–	–

^a The ion current generated depends on the characteristics of the compound concerned and it is not a true quantitation.

Table III. Antibacterial and antifungal activity of propolis samples (extracts with 70% ethanol).

Sample	<i>S. aureus</i>	<i>E. coli</i>	<i>C. albicans</i>
	(diameter of the inhibitory zone \pm stand. deviation, mm) ^a		
LOV	20 \pm 1	0	13 \pm 1
BUR	20 \pm 1	0	13.7 \pm 0.6
GR	18.7 \pm 0.6	0	12 \pm 0
BU-7	19 \pm 2	12 \pm 1	14 \pm 1
BU-8	20 \pm 2	12 \pm 0	16 \pm 1
MU	21 \pm 2	13.3 \pm 0.6	17 \pm 2
IZ	19 \pm 2	12 \pm 1	13 \pm 1
AN	21.7 \pm 0.6	14 \pm 0	13 \pm 1
AL-1	19 \pm 1	0	0
AL-2	18.7 \pm 10.6	0	12 \pm 0

^a Mean of three measurements.

the main “poplar” phenolics. They confirm also that in border areas, such as Algeria, where poplars are not always available for propolis collection, other plant sources are used but this does not affect the antibacterial properties of bee glue.

Acknowledgements

The authors wish to thank Miss N. Nikolova for technical assistance. Partial support of this work by the National Council for Scientific Research (Contract #X-715) is gratefully acknowledged.

- Bankova V., de Castro S. L. and Marcucci, M. C. (2000), Propolis: recent advances in chemistry and plant origin. *Apidologie* **31**, 3–15.
- Bankova V., Dylgerov A., Popov S., Evstatieva L., Kuleva L., Pureb O. and Zamjansan Z. (1992), Propolis produced in Bulgaria and Mongolia: phenolic composition and plant origin. *Apidologie* **23**, 79–85.
- Bankova V., Marcucci M. C., Simova S., Nikolova N., Kujumgiev A. and Popov S. (1996), Antibacterial diterpenic acids from Brazilian propolis. *Z. Naturforsch.* **51c**, 277–280.
- Banskota A. H., Tezuka Y., Prasian J. K., Matsushige K., Saiki I. and Kadota Sh. (1998), Chemical constituents of Brazilian propolis and their cytotoxic activities. *J. Nat. Prod.* **61**, 896–900.
- Burdock G. A. (1998), Review of the biological properties and toxicity of bee propolis (propolis). *Food Chem. Toxicol.* **36**, 347–363.
- Christov R., Bankova V., Hegazi A., Abd El Hady F. and Popov S. (1998), Chemical composition of Egyptian propolis. *Z. Naturforsch.* **53c**, 197–200.
- Greenaway W., Scaysbrook T. and Whatley F. R. (1990), The composition and plant origins of propolis: a report of work at Oxford. *Bee World* **71**, 107–118.
- Kujumgiev A., Tsvetkova I., Serkedjieva Yu., Bankova V., Christov R. and Popov S. (1999), Antibacterial, antifungal and antiviral activity of propolis of different geographic origin. *J. Ethnopharm.* **64**, 235–240.
- Marcucci M. C. (1995), Propolis: chemical composition, biological properties and therapeutical activity. *Apidologie* **26**, 83–99.
- Martos L., Cossentini M., Ferreres F. and Tomas-Barberan F. A. (1997), Flavonoid composition of Tunisian honey and propolis. *J. Agric. Food Chem.* **54**, 2824–2829.
- Spooner F. D. and Sykes G. (1972), Laboratory assessment of antibacterial activity. In: *Methods in Microbiology*, Vol. **7B** (Norris J. R., Ribbons D. W., eds.). Academic Press, London, pp. 216–217.
- Wollenweber E. and Buchmann S. L. (1997), Feral honey bees in the Sonoran desert: propolis sources other than poplar (*Populus* spp.). *Z. Naturforsch.* **52c**, 530–535.